

LAMPIRAN

Lampiran 1 Pembuatan Ekstrak Bunga Telang

1. Perhitungan Persentase Bobot Kering Terhadap Bobot Basah Sampel

Berat bunga telang basah = 850.00 gram

Berat bunga telang kering = 247,13 gram

$$= \frac{\text{berat kering}}{\text{berat basah}} \times 100\%$$

$$= \frac{247,13 \text{ gram}}{850.00 \text{ gram}} \times 100\%$$

$$= 29,07\%$$

Lampiran 2 Perhitungan Rendemen Ekstrak Maserasi Bunga Telang

1. Perhitungan Ekstrak

$$\begin{aligned}
 \text{Berat sampel} &= 50 \text{ gram} & (\text{x}) \\
 \text{Berat cawan kosong} &= 75,84 \text{ gram} & (\text{b}) \\
 \text{Berat cawan+isi} &= 85,41 \text{ gram} & (\text{c}) \\
 \text{Berat ekstrak} &= \text{c} - \text{b} \\
 &= 85,41 \text{ gram} - 75,84 \text{ gram} \\
 &= 9,57 \text{ gram} & (\text{y})
 \end{aligned}$$

2. Hasil Rendemen

$$\begin{aligned}
 \% \text{ Rendemen} &= \frac{y}{x} \times 100\% \\
 &= \frac{9,57 \text{ gram}}{50 \text{ gram}} \times 100\% \\
 &= 19,14\% \text{ b/b}
 \end{aligned}$$

Jadi, hasil ekstraksi menggunakan metode maserasi dari berat sampel awal 50 gram memperoleh ekstrak kental sebanyak 9,57 gram dengan rendemen sebesar 19,14% b/b.

Lampiran 3 Perhitungan rendemen Ekstrak Perkolasi Bunga Telang

1. Perhitungan Ekstrak

$$\begin{aligned}
 \text{Berat sampel} &= 50 \text{ gram} \quad (\text{x}) \\
 \text{Berat cawan kosong} &= 54,68 \text{ gram} \quad (\text{b}) \\
 \text{Berat cawan + isi} &= 73,14 \text{ gram} \quad (\text{c}) \\
 \text{Berat ekstrak} &= \text{c} - \text{b} \\
 &= 73,14 \text{ gram} - 54,68 \text{ gram} \\
 &= 18,46 \text{ gram} \quad (\text{y})
 \end{aligned}$$

2. Hasil Rendemen

$$\begin{aligned}
 \% \text{ Rendemen} &= \frac{y}{x} \times 100\% \\
 &= \frac{18,46 \text{ gram}}{50 \text{ gram}} \times 100\% \\
 &= 36,92\% \text{ b/b}
 \end{aligned}$$

Jadi, hasil ekstraksi menggunakan metode perkolasi dari berat sampel awal 50 gram memperoleh ekstrak kental sebanyak 18,46 gram dengan rendemen sebesar 36,92% b/b.

Lampiran 4 Perhitungan Rf dan hRf Kromatografi Lapis Tipis

1. Ekstrak Maserasi Bunga Telang

a. Noda 1

$$Rf = \frac{\text{jarak tempuh sampel}}{\text{jarak tempuh pelarut}}$$

$$= \frac{7,3 \text{ cm}}{7,8 \text{ cm}}$$

$$= 0,935$$

$$hRf = \frac{\text{jarak tempuh sampel}}{\text{jarak tempuh pelarut}} \times 100$$

$$= \frac{7,3 \text{ cm}}{7,8 \text{ cm}} \times 100$$

$$= 93,5$$

2. Ekstrak Perkolasi Bunga Telang

a. Noda 1

$$Rf = \frac{\text{jarak tempuh sampel}}{\text{jarak tempuh pelarut}}$$

$$= \frac{7 \text{ cm}}{7,8 \text{ cm}}$$

$$= 0,897$$

$$hRf = \frac{\text{jarak tempuh sampel}}{\text{jarak tempuh pelarut}} \times 100$$

$$= \frac{7 \text{ cm}}{7,8 \text{ cm}} \times 100$$

$$= 89,7$$

Lampiran 5 Perhitungan Pembuatan Larutan Uji**1. Pembuatan Larutan DPPH 50 ppm**

$$\begin{aligned} \text{DPPH} &= 50 \text{ ppm} \\ &= 50 \mu\text{g/ml} \\ &= 0,05 \text{ mg/ml} \\ \text{DPPH yang dibutuhkan} &= 0,05 \text{ mg/ml} \times 100 \text{ ml} \\ &= 5 \text{ mg} \\ \text{Methanol ad} &= 100 \text{ ml} \end{aligned}$$

2. Pembuatan Larutan Induk 1000 ppm

$$\begin{aligned} &= 1000 \text{ ppm} \\ &= 1000 \mu\text{g/ml} \\ &= 1 \text{ mg/ml} \\ \text{Ekstrak yang dibutuhkan} &= 1 \text{ mg/ml} \times 50 \text{ ml} \\ &= 50 \text{ mg} \\ \text{Methanol ad} &= 50 \text{ ml} \end{aligned}$$

3. Pembuatan Larutan Seri 20 ppm, 40 ppm, 50 ppm, Dan 80 ppm

$$\begin{aligned}
 20 \text{ ppm} & V_1 \times N_1 = V_2 \times N_2 \\
 V_1 \times 1000 \text{ ppm} & = 10 \text{ ml} \times 20 \text{ ppm} \\
 V_1 & = \frac{200}{1000} \text{ ml} \\
 & = 0,2 \text{ ml} \longrightarrow \text{methanol ad 10 ml}
 \end{aligned}$$

$$\begin{aligned}
 40 \text{ ppm} & V_1 \times N_1 = V_2 \times N_2 \\
 V_1 \times 1000 \text{ ppm} & = 10 \text{ ml} \times 40 \text{ ppm} \\
 V_1 & = \frac{400}{1000} \text{ ml} \\
 & = 0,4 \text{ ml} \longrightarrow \text{methanol ad 10 ml}
 \end{aligned}$$

$$\begin{aligned}
 50 \text{ ppm} & V_1 \times N_1 = V_2 \times N_2 \\
 V_1 \times 1000 \text{ ppm} & = 10 \text{ ml} \times 50 \text{ ppm} \\
 V_1 & = \frac{500}{1000} \text{ ml} \\
 & = 0,5 \text{ ml} \longrightarrow \text{methanol ad 10 ml}
 \end{aligned}$$

$$\begin{aligned}
 80 \text{ ppm} & V_1 \times N_1 = V_2 \times N_2 \\
 V_1 \times 1000 \text{ ppm} & = 10 \text{ ml} \times 80 \text{ ppm} \\
 V_1 & = \frac{800}{1000} \text{ ml} \\
 & = 0,8 \text{ ml} \longrightarrow \text{methanol ad 10 ml}
 \end{aligned}$$

4. Pembuatan Larutan Induk Control Vitamin C 1000 ppm

Vitamin	= 1000 ppm
	= 1000 µg/ml
	= 1 mg/ml
Ekstrak yang dibutuhkan	= 1 mg/ml x 100 ml
	= 100 mg
Methanol ad	= 100 ml

5. Pembuatan Larutan Seri Vitamin C 20 ppm, 40 ppm, 50 ppm Dan 80 ppm

$$\begin{aligned}
 & 20 \text{ ppm} \quad V_1 \times N_1 = V_2 \times N_2 \\
 & V_1 \times 1000 \text{ ppm} = 10 \text{ ml} \times 20 \text{ ppm} \\
 & V_1 = \frac{200}{1000} \text{ ml} \\
 & = 0,2 \text{ ml} \longrightarrow \text{methanol ad 10 ml} \\
 & 40 \text{ ppm} \quad V_1 \times N_1 = V_2 \times N_2 \\
 & V_1 \times 1000 \text{ ppm} = 10 \text{ ml} \times 40 \text{ ppm} \\
 & V_1 = \frac{400}{1000} \text{ ml} \\
 & = 0,4 \text{ ml} \longrightarrow \text{methanol ad 10 ml} \\
 & 50 \text{ ppm} \quad V_1 \times N_1 = V_2 \times N_2 \\
 & V_1 \times 1000 \text{ ppm} = 10 \text{ ml} \times 50 \text{ ppm} \\
 & V_1 = \frac{500}{1000} \text{ ml} \\
 & = 0,5 \text{ ml} \longrightarrow \text{methanol ad 10 ml} \\
 & 80 \text{ ppm} \quad V_1 \times N_1 = V_2 \times N_2 \\
 & V_1 \times 1000 \text{ ppm} = 10 \text{ ml} \times 80 \text{ ppm} \\
 & V_1 = \frac{800}{1000} \text{ ml} \\
 & = 0,8 \text{ ml} \longrightarrow \text{methanol ad 10 ml}
 \end{aligned}$$

Lampiran 6 Perhitungan % Inhibisi Ekstrak Maserasi Bunga Telang

1. Larutan Blanko

Membuat larutan blanko dengan menambahkan 1,5 ml methanol dan 3 ml DPPH

Replikasi	Data Absorbansi
1	0,525
2	0,525
3	0,525
Rata - Rata	0,525

2. Perhitungan % Inhibisi Ekstrak Maserasi Bunga Telang

$$20 \text{ ppm} = \frac{(\text{Rata-rata abs.blanko}) - (\text{Rata-rata abs.sampel})}{(\text{Rata-rata abs.blanko})} \times 100\%$$

$$= \frac{0,525 - 0,456}{0,525} \times 100\%$$

$$= 13,14\%$$

$$40 \text{ ppm} = \frac{(\text{Rata-rata abs.blanko}) - (\text{Rata-rata abs.sampel})}{(\text{Rata-rata abs.blanko})} \times 100\%$$

$$= \frac{0,525 - 0,448}{0,525} \times 100\%$$

$$= 14,66\%$$

$$50 \text{ ppm} = \frac{(\text{Rata-rata abs.blanko}) - (\text{Rata-rata abs.sampel})}{(\text{Rata-rata abs.blanko})} \times 100\%$$

$$= \frac{0,525 - 0,438}{0,525} \times 100\%$$

$$= 16,57\%$$

$$80 \text{ ppm} = \frac{(\text{Rata-rata abs.blanko}) - (\text{Rata-rata abs.sampel})}{(\text{Rata-rata abs.blanko})} \times 100\%$$

$$= \frac{0,525 - 0,436}{0,525} \times 100\%$$

$$= 16,95\%$$

Lampiran 7 Perhitungan % Inhibisi Ekstrak Perkolasi Bunga Telang

1. Larutan Blanko

Membuat larutan blanko dengan menambahkan 1,5 ml methanol dan 3 ml DPPH

Replikasi	Data Absorbansi
1	0,525
2	0,525
3	0,525
Rata - Rata	0,525

2. Perhitungan % Inhibisi Ekstrak Perkolasi Bunga Telang

$$20 \text{ ppm} = \frac{(\text{Rata-rata abs.blanko}) - (\text{Rata-rata abs.sampel})}{(\text{Rata-rata abs.blanko})} \times 100\%$$

$$= \frac{0,525 - 0,396}{0,525} \times 100\%$$

$$= 24,57\%$$

$$40 \text{ ppm} = \frac{(\text{Rata-rata abs.blanko}) - (\text{Rata-rata abs.sampel})}{(\text{Rata-rata abs.blanko})} \times 100\%$$

$$= \frac{0,525 - 0,373}{0,525} \times 100\%$$

$$= 28,95\%$$

$$50 \text{ ppm} = \frac{(\text{Rata-rata abs.blanko}) - (\text{Rata-rata abs.sampel})}{(\text{Rata-rata abs.blanko})} \times 100\%$$

$$= \frac{0,525 - 0,343}{0,525} \times 100\%$$

$$= 34,66\%$$

$$80 \text{ ppm} = \frac{(\text{Rata-rata abs.blanko}) - (\text{Rata-rata abs.sampel})}{(\text{Rata-rata abs.blanko})} \times 100\%$$

$$= \frac{0,525 - 0,198}{0,525} \times 100\%$$

$$= 62,28\%$$

Lampiran 8 Perhitungan % Inhibisi Kontrol Positif Vitamin C

1. Larutan Blanko

Replikasi	Data Absorbansi
1	0,525
2	0,525
3	0,525
Rata - Rata	0,525

2. Perhitungan % Inhibisi Kontrol Positif Vitamin C

$$20 \text{ ppm} = \frac{(\text{Rata-rata abs.blanko}) - (\text{Rata-rata abs.sampel})}{(\text{Rata-rata abs.blanko})} \times 100\%$$

$$= \frac{0,525 - 0,047}{0,525} \times 100\%$$

$$= 91,04\%$$

$$40 \text{ ppm} = \frac{(\text{Rata-rata abs.blanko}) - (\text{Rata-rata abs.sampel})}{(\text{Rata-rata abs.blanko})} \times 100\%$$

$$= \frac{0,525 - 0,041}{0,525} \times 100\%$$

$$= 92,19\%$$

$$50 \text{ ppm} = \frac{(\text{Rata-rata abs.blanko}) - (\text{Rata-rata abs.sampel})}{(\text{Rata-rata abs.blanko})} \times 100\%$$

$$= \frac{0,525 - 0,030}{0,525} \times 100\%$$

$$= 94,28\%$$

$$80 \text{ ppm} = \frac{(\text{Rata-rata abs.blanko}) - (\text{Rata-rata abs.sampel})}{(\text{Rata-rata abs.blanko})} \times 100\%$$

$$= \frac{0,525 - 0,027}{0,525} \times 100\%$$

$$= 94,85\%$$

Lampiran 9 Perhitungan Nilai IC₅₀

1. Ekstrak Maserasi Bunga Telang

$$\text{Hasil Kurva} \quad y = 0,1348x - 0,4408$$

$$R^2 = 0,907$$

$$\text{Rumus IC}_{50} \quad y = ax + b$$

$$5 = 0,1348x + 0,4408$$

$$x = \frac{5 - 0,4408}{0,1348}$$

$$= 40,36201$$

$$IC_{50} = \text{antilog } 40,36201$$

$$= 317,49$$

2. Ekstrak Perkolasi Bunga Telang

$$\text{Hasil Kurva} \quad y = 0,0128x + 1,1436$$

$$R^2 = 0,7542$$

$$\text{Rumus IC}_{50} \quad y = ax + b$$

$$5 = 0,0128x + 1,1436$$

$$x = \frac{5 - 1,1436}{0,0128}$$

$$= 301,28125$$

$$IC_{50} = \text{antilog } 301,28125$$

$$= 6,31$$

3. Vitamin C

$$\text{Hasil Kurva} \quad y = 0,1324x - 10,699$$

$$R^2 = 0,8857$$

$$\text{Rumus IC}_{50} \quad y = ax + b$$

$$5 = 0,1324x + 10,699$$

$$x = \frac{5 - 10,699}{0,1324}$$

$$= 118,57250$$

$$IC_{50} = \text{antilog } 118,57250$$

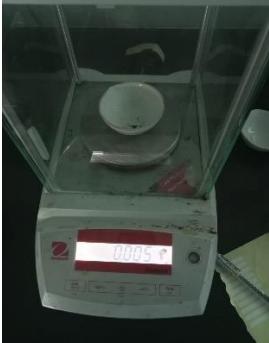
$$= 0,25$$

Lampiran 10 Dokumentasi Penelitian

No	Gambar	Keterangan
1.		Sampel bunga telang
2.		Serbuk simplisia bunga telang
3.	A photograph showing two stainless steel containers with lids, each covered by a black plastic bag, resting on a dark, reflective countertop in what appears to be a laboratory or kitchen setting. In the background, there are shelves with various bottles and equipment.	Maserasi selama 1 x 24 jam di tempat yang minim cahaya dan perkolasai selama 1 jam

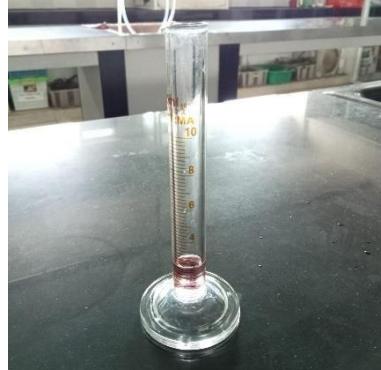
No	Gambar	Keterangan
4.		Penguapan ekstrak menggunakan <i>waterbath</i>
5.		Uji flavonoid

Lampiran 11 Dokumentasi Uji Aktivitas Antioksidan

No	Gambar	Keterangan
1.		Menimbang Vitamin C
2.		Larutan induk Vitamin C
3.		Menimbang serbuk DPPH

No	Gambar	Keterangan
4.		Larutan DPPH
5.		Menimbang ekstrak kental bunga telang maserasi
6.		Larutan induk ekstraksi maserasi

No	Gambar	Keterangan
7.		Menimbang ekstrak kental bunga telang perkolasai
8.		Larutan induk ekstraksi perkolasai
9.		Larutan seri ekstrak maserasi bunga telang + metanol

No	Gambar	Keterangan
10.		Larutan seri ekstrak perkolasii bunga telang + metanol
11.		Larutan seri vitamin c + metanol
12.		Blanko DPPH

No	Gambar	Keterangan
13.		Proses inkubasi

CURRICULUM VITAE



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PENDIDIKAN

SD : MI Nurul Ulum Pegiringan
MTS : MTs Nurul Ulum Pegiringan
SMK : SMK Kesehatan Amanah Husada Pemalang
D III : Politeknik Harapan Bersama Tegal
Judul TA : Pengaruh Perbedaan Metode Ekstraksi Terhadap Nilai IC₅₀ Ekstrak Bunga Telang (*Clitoria ternatea* L.)

NAMA ORANG TUA

Ayah : Abdul Mukti
Ibu : Titin Supriyatin

ALAMAT ORANG TUA

Ayah : Desa Pegiringan Rt 002/Rw 005 Kec.Bantarbolang Kab. Pemalang
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